

Fe-S-Zn (Iron-Sulfur-Zinc)

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The previous review of this system by [1988Rag1] presented a schematic liquidus projection, a reaction scheme and four isothermal sections at 850, 700, 600, and 400 °C, mainly from the studies of [1966Bar]. An update by [2004Rag] gave an additional isothermal section at 927 °C. Recently, [2007Tan] determined an isothermal section for this system at 450 °C.

Binary Systems

The Fe-S system [1988Rag2] has two intermediate phases at 450 °C: Fe_{1-x}S (NiAs-type hexagonal) is stable between 50 and 55 at.% S; FeS_2 (C2, pyrite-type cubic) forms peritectically at 743 °C and transforms to FeS_2 (C18, marcasite-type orthorhombic) at 425 °C. The intermediate phases in the Fe-Zn system are: Γ ($\text{Fe}_3\text{Zn}_{10}$; Cu_5Zn_8 -type cubic), Γ_1 ($\text{Fe}_{11}\text{Zn}_{40}$; cubic, space group $F43m$, 408 atoms/cell), δ (FeZn_{10} ; FeZn_{10} -type hexagonal), and ζ (CoZn_{13} -type monoclinic). In the Zn-S system, a congruently-melting compound ZnS (B4, wurtzite-type hexagonal) transforms to ZnS (B3, sphalerite-type cubic) at 1020 °C during cooling.

Ternary Isothermal Section

With starting metal powders of >99.99 % purity, [2007Tan] mixed and compacted powder samples, which were annealed in evacuated quartz tubes at 450 °C for 2 months and quenched in water. The phase equilibria were studied with x-ray powder diffraction, and a scanning electron microscope equipped with energy dispersive x-ray spectroscope. The isothermal section at 450 °C constructed by [2007Tan] is shown in Fig. 1. It is very similar to the section at 400 °C given by [1988Rag1], except for the presence of the Zn-rich liquid at 450 °C in place of (Zn). The solubility of S in Fe-Zn compounds and Zn in Fe-S compounds was found to be negligible.

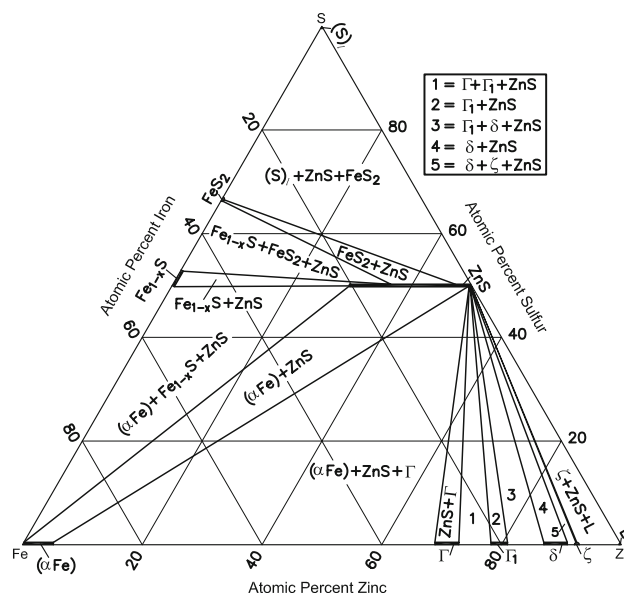


Fig. 1 Fe-S-Zn isothermal section at 450 °C [2007Tan]

References

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